

STRAND BREAKS IN PLASMID DNA FOLLOWING POSITIONAL CHANGES OF AUGER-ELECTRON-EMITTING RADIONUCLIDES

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The purpose of our studies is to elucidate the kinetics of DNA strand breaks caused by low-energy Auger electron emitters in close proximity to DNA. Previously we have studied the DNA break yields in plasmids after the decay of indium-111 bound to DNA or free in solution. In this work, we compare the DNA break yields in supercoiled DNA of iodine-125 decaying close to DNA following DNA intercalation, minor-groove binding, or surface binding, and at a distance from DNA. Supercoiled DNA, stored at 4°C to accumulate radiation dose from the decay of ^{125}I , was then resolved by gel electrophoresis into supercoiled, nicked circular, and linear forms, representing undamaged DNA, single-strand breaks, and double-strand breaks respectively. DNA-intercalated or groove-bound ^{125}I is more effective than surface-bound radionuclide or ^{125}I free in solution. The hydroxyl radical scavenger DMSO protects against damage by ^{125}I free in solution but has minimal effect on damage by groove-bound ^{125}I .

There is general agreement that under certain circumstances Auger-electron-emitting radionuclides can be extremely radiotoxic and produce extensive DNA damage, including the shattering of both strands of duplex DNA and of poly- and oligonucleotides. The degree of damage appears to depend upon the location of the decaying atom and microdosimetric calculations predict such. For example, the cytoplasmic dye iodorhodamine, labeled with ^{125}I , produces in Chinese hamster cells a survival curve that resembles low-LET irradiation (1). When covalently bound to DNA, ^{125}I produces a high-LET curve under the same experimental circumstances (2), while intercalated ^{125}I produces a high-LET curve with RBE 50 to 60% that of the incorporated radionuclide (3). We have also found that ^{123}I incorporated into mammalian cell DNA produces

about 0.7 double-strand breaks (dsb) per decay (4) in keeping with theory and the original observation that each decay of ^{125}I produces one dsb (5).

In molecular systems, Martin & Haseltine (6) have shown that covalently linked ^{125}I shatters both ipsilateral and contralateral strands of DNA in duplex oligonucleotides. Martin and his colleagues have further determined that the aminoacridine iodorivanol, a DNA intercalator, when labeled with ^{125}I , generates dsb in DNA (7). Linz & Stocklin (8) have also demonstrated that ^{125}I covalently bound to plasmid DNA is more efficient than ^{125}I free in solution at causing dsb, while we have shown that ^{111}In in its ionic form, bound to the outer side of DNA through the phosphate and nitrogenous base groups of the nucleotides (9), causes the loss of supercoiled pBR322 and the production of single-strand breaks (ssb) with 10 to 20 times the efficiency of the isotope kept free in solution by coordination with DTPA (10). Panyutin & Neumann (11) have recently reported that an ^{125}I -labeled, specific triplex-forming oligonucleotide, binding to the major groove, produces dsb with an efficiency of 0.8.

Four mechanisms for the production of dsb have been proposed (12): double-strand breaks from two nearby single-strand breaks (ssb) on opposite strands (the quadratic two-event mechanism); dsb from radical transfer

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Table 1*Positional decay of Auger-electron-emitting radio-nuclides*

Type	Example
Minor groove binding	¹²⁵ I-Hoechst 33342
Intercalation	2- ¹²⁵ I]iodoacridine
Outside binding	4- ¹²⁵ I]iodoacridine
Free	[¹²⁵ I]iodoantipyrine

between strands (Siddiqi-Bothe hypothesis, (13)); dsb from clusters of watery OH radicals from a single event (Ward's multiply damaged site hypothesis, (14)); and direct action on both strands by a single densely ionizing event. It has been postulated that scavenging molecules will protect against the first two but not the last two mechanisms.

In order to explore the dependence of DNA damage on the position of the decaying Auger-electron-emitting atom we have employed two well-characterized plasmids, pBR322 and pUC19. In their native form these plasmids consist of supercoiled DNA; a ssb converts them to a circular (nicked) form and a dsb to a linear form. The three forms are easily separated by electrophoresis on agarose gels. In our studies, a number of chemical entities have been used to explore radiolytic effects, i.e. damage to plasmid DNA. The radiochemicals include: iodinated Hoechst dye 33342 (H33342) for groove-bound ¹²⁵I; 2-iodoacridine for intercalated ¹²⁵I; 4-iodoacridine for non-specifically bound ¹²⁵I; iodoantipyrine for ¹²⁵I free in solution; (Table 1). In two of these systems (¹²⁵I-H33342 and [¹²⁵I]iodoantipyrine), the radioprotective effects of DMSO were also studied.

Material and Methods

Preparation of plasmids

The pBR322 plasmid was grown in *E. coli* (recA, thi⁻) bacteria in L.B. medium containing 100 µg/ml ampicillin at 37°C and the pUC19 plasmid in *E. coli* (DH5 α) in T.B. medium containing ampicillin (150 µg/ml) and thiamine (10 µg/ml). The plasmids were amplified (pBR322 for 18 h in the presence of 5 µCi/ml ³H-thymidine and pUC19 for 42 h in the presence of 36 µCi/ml ³H-thymidine) and then purified on a commercially available resin column (Qiagen Incorporated, Studio City, CA).

Preparation of radiochemicals

2-Iodoacridine and 4-iodoacridine were synthesized as previously described (15). [¹²⁵I]iodoantipyrine was purchased from ICN Pharmaceuticals, Incorporated (Costa Mesa, CA). The method for iodinating Hoechst dye 33342 will be published in the near future.

Experimental protocols

The individual protocols are summarized in Table 2. pBR322 and pUC19 plasmids (100 µg) prepared as described above were placed in solution (final volume = 0.1 ml) containing the radiochemicals listed in Table 1, and the samples were stored at 4°C to accumulate radiation dose (1–60 days). The ¹²⁵I-labeled iodoacridines along with 0.2% acetone were dissolved in 10 mM sodium phosphate buffer, pH 4.9, containing 0.1 mM EDTA and mixed with pBR322 plasmids that were extensively dialyzed in the same buffer immediately prior to use. When ¹²⁵I-H33342 and [¹²⁵I]-iodoantipyrine were compared, pUC19 plasmids were mixed with each of these agents in phosphate-buffered saline, pH 7.4, and in some instances DMSO (0.2 M) was added to assess the effects of hydroxyl radical scavenging.

DNA strand-break assay

After the accumulation of the desired radiation dose from the decay of the radioisotope, the plasmid DNA samples were resolved by gel electrophoresis into supercoiled, nicked circular, and linear forms, representing undamaged DNA, ssb, and dsb respectively. Distinct bands of DNA formed when one microgram of DNA from each sample was applied on 0.8% agarose gel and the electrophoresis apparatus was run for 18 h (3 volts/cm) in the dark in TBE buffer (89 mM tris, 89 mM borate, 1 mM EDTA), pH 8.0, containing 0.5 µg/ml ethidium bromide. The agarose gels were washed in phosphate buffer, photographed (under UV), and the DNA-containing bands excised under UV light and dissolved in 0.5 ml of 1 M HCl by heating at 60°C for 15 min. In the radiolabeled acridine studies, the concentration of ³H-DNA in the DNA bands, which is proportional to the amount of DNA present, was determined by assaying the radioactivity in a liquid scintillation counter. Wherever needed, the counts were corrected for ¹²⁵I spill over. In the ¹²⁵I-iodoantipyrine and ¹²⁵I-labeled H33342 studies, the DNA content within each band was estimated visually by two independent observers.

Table 2*Experimental protocols*

pBR322		pUC19 (± DMSO)		
Intercalated 2- ¹²⁵ I]iodoacridine	versus	Bound 4- ¹²⁵ I]iodoacridine	Groove-bound ¹²⁵ I-Hoechst 33342	versus Free [¹²⁵ I]iodoantipyrine

Results and Discussion

The experiments reported herein address the production of single- and double-strand breaks under various circumstances.

DNA-bound and intercalated ¹²⁵I

Acridines and proflavines, flat aromatic compounds, interact with DNA by sandwiching themselves between the stacked bases at right angles to the long axis of the helix (16). If a bulky iodine atom is attached to the leading edge of such molecules, as with 4-iodoacridine, the planar molecule cannot enter the stack; if the iodine is attached to the trailing edge, as with 2-iodoacridine, intercalation can proceed. 4-Iodoacridine when mixed with DNA shows a small hypochromic shift but no bathochromic one, signifying binding but not intercalation, while 2-iodoacridine demonstrates a bathochromic shift with an isosbestic point indicating that intercalation has taken place (15). Incubation of the two ¹²⁵I-labeled compounds with plasmid DNA in the presence of acetone, an hydroxyl radical scavenger, shows efficient dsb formation with the 2-iodo compound (0.67 dsb on average per disintegration) and much less dsb production with the 4-iodo compound (0.27 dsb on average per disintegration) (Table 3). The ssb/dsb ratio changes from 0.6 with 2-iodoacridine to approximately 16 with 4-iodoacridine.

Groove-bound and free ¹²⁵I

The bis-benzimidazole dyes, Hoechst 33258 and 33342, have been shown to bind to the minor groove of DNA in A-T rich regions (17). A radioiodinated analog of Hoechst 33342 has recently been synthesized and characterized. It has been shown to concentrate in the nuclei of mammalian cells (to be published). When the dyes are iodinated and mixed with DNA, the iodo group is located between the two strands of DNA in the minor groove (18). Iodination does not change the binding constant or the maximum number of molecules bound to DNA (Table 4). The DNA of plasmids exposed to ¹²⁵I-labeled Hoechst 33342 has been compared with that exposed to [¹²⁵I]iodoantipyrine, a diffusible, water-soluble compound. Inspection of the gels from plasmid-iodoantipyrine incubation mixtures reveals a rapid conversion of supercoiled to nicked (ssb) DNA with the progressive appearance of linear DNA (dsb) (Fig.

Table 4

Characteristics of groove-binding Hoechst compounds

Ligand	K _a (× 10 ⁷ M ⁻¹)	Ligand/nucleotide
Hoechst 33258	2.64	0.0275
Hoechst 33342	2.84	0.0282
¹²⁵ I-H 33342	2.57	0.0280

1). Inspection of the gels from plasmid-iodinated Hoechst 33342 incubation mixtures shows the simultaneous appearance of the nicked and linear forms, indicating the direct formation of dsb (Fig. 1). Semiquantitative visual analysis reiterates these findings and demonstrates that after an equal number of decays, about five times as many dsb are produced in DNA by the radioiodinated groove binder as by [¹²⁵I]iodoantipyrine (Fig. 2). Furthermore, addition of the hydroxyl radical scavenger DMSO leads to a substantial reduction in the rate of ssb production by iodoantipyrine and virtually no dsb production, while DMSO has little effect on either ssb or dsb production by the groove

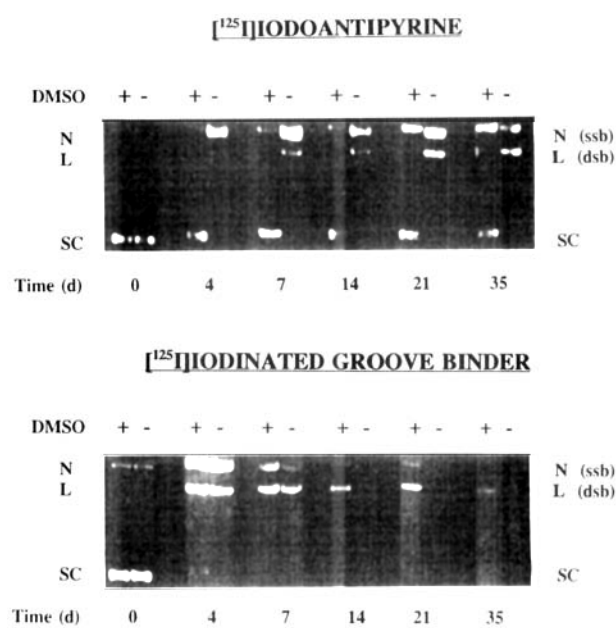


Fig. 1. Agarose gels of pUC19 supercoiled plasmid DNA exposed over time to ¹²⁵I-labeled Hoechst 33342 and [¹²⁵I]iodoantipyrine with and without DMSO; SC = supercoiled, N = nicked (ssb), L = linear (dsb). Note disappearance of total plasmid DNA from gel over time course of experiment.

Table 3

DNA strand breaks per disintegration of ¹²⁵I-labeled iodoacridines

DNA	2-[¹²⁵ I]iodoacridine		4-[¹²⁵ I]iodoacridine	
	ssb	dsb	ssb	dsb
ssb	0.38		4.33	
dsb	0.67		0.27	
				2-[¹²⁵ I]iodoacridine
				4-[¹²⁵ I]iodoacridine
				ssb
				dsb

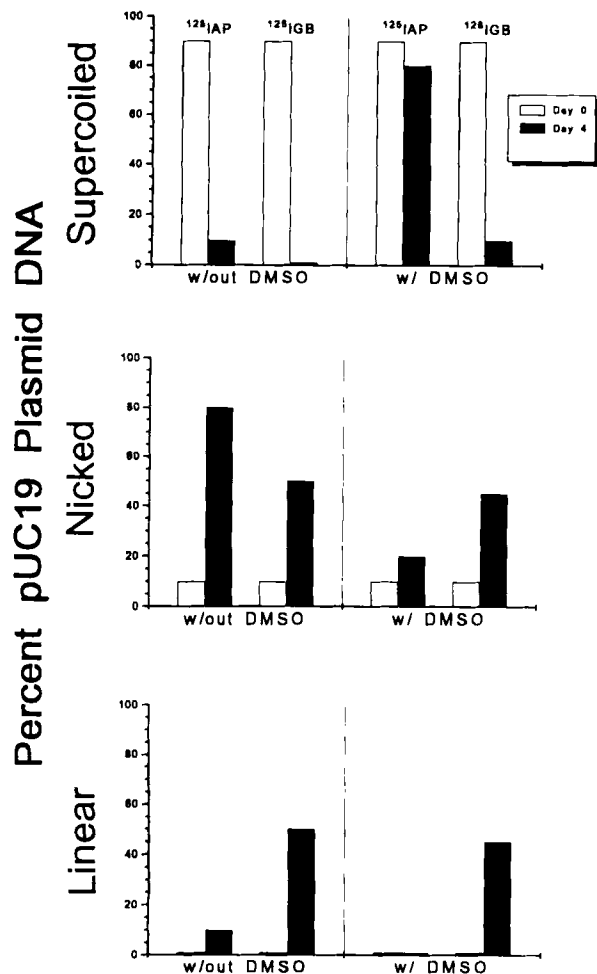


Fig. 2. Semiquantitative analysis of gels (days 0 and $4 \pm$ DMSO) in Figure 1. Percentages take into account only three major species of DNA (supercoiled, nicked, and linear) and not other plasmid breakdown products.

binder (Figs. 2 and 3). This observation is consistent with dsb production that is caused by radical clusters or direct action by the ^{125}I affixed to the groove binder, i.e. in close proximity to DNA.

Conclusions

In plasmid DNA, damage caused by Auger electron emitters depends strongly on the position of the decaying radionuclide relative to the supercoiled structure. Our observations include the following:

- Ionic ^{111}In , bound on the surface of DNA through the phosphate and nitrogenous base groups of nucleotides, destroys the supercoiled form of pBR322 and produces ssb with 10 to 20 times the efficiency of the isotope not bound and kept free in solution by coordination with DTPA (10).
- Intercalated ^{125}I (2- ^{125}I iodoacridine) is two to three times as effective as bound but unintercalated ^{125}I (4- ^{125}I iodoacridine) in producing dsb (15).

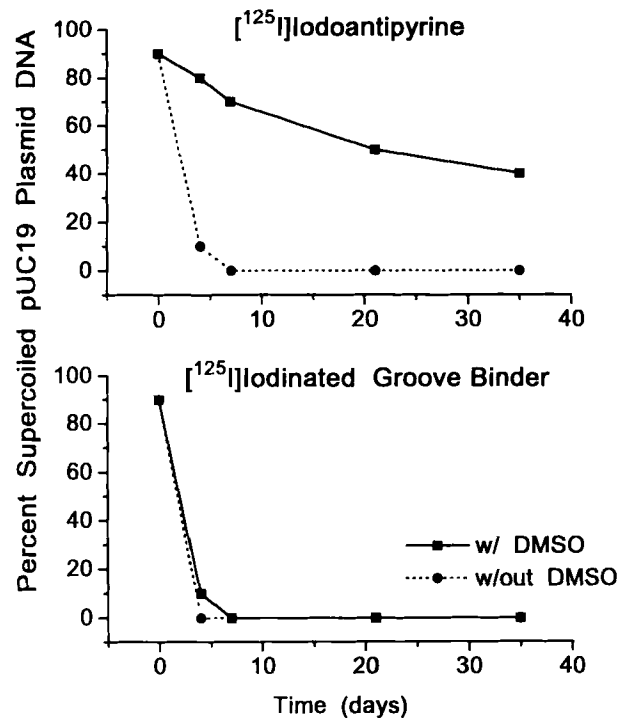


Fig. 3. Percentage of supercoiled DNA remaining over time after exposure of pUC19 plasmid DNA to ^{125}I -labeled Hoechst 33342 and ^{125}I iodoantipyrene with and without DMSO (data generated from Fig. 1). Percentages take into account only three major species of DNA (supercoiled, nicked, and linear) and not other plasmid breakdown products.

- Groove-bound ^{125}I (radioiodinated Hoechst 33342) is five to seven times more efficient than free ^{125}I (^{125}I iodoantipyrene) in producing dsb.
- The hydroxyl radical scavenger DMSO protects against damage by free ^{125}I (^{125}I iodoantipyrene) but has little influence on damage produced by groove-bound ^{125}I (radioiodinated Hoechst 33342).

These observations are consistent with microdosimetric analyses and the production of dsb by a single-event mechanism related to the high radical density produced by the decaying Auger-electron-emitting atom when in close proximity to DNA as found in intercalation and minor-groove binding. We predict the same when ^{125}I is covalently bound.

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